# Chen Northern, Inc.

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MO. BURNA, M.

# WASTE MATERIAL SAMPLING REPORT NELLIE GRANT MINE JEFFERSON COUNTY, MONTANA

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WASTE MATERIAL SAMPLING REPORT

NELLIE GRANT MINE

JEFFERSON COUNTY, MONTANA

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### **DRAFT**

# WASTE MATERIAL SAMPLING REPORT NELLIE GRANT MINE JEFFERSON COUNTY, MONTANA

## Prepared for:

Mr. Stu Levit
Abandoned Mine Reclamation Bureau
Montana Department of State Lands
1625 Eleventh Avenue
Helena, Montana 59620

Prepared by:

Chen-Northern, Inc. P.O. Box 4699 Helena, Montana

October, 1990

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## TABLE OF CONTENTS

		<u>Pa</u>	age
1.0	INTRODUCTION		1
2.0	<ul><li>2.1 SITE RECONNAISSANCE</li><li>2.2 SITE SAFETY PLAN AN</li><li>2.3 SAMPLING AND ANALY</li><li>2.4 WASTE MATERIAL</li></ul>	E	
3.0	3.1 WASTE MATERIAL IDE		10 10 13
4.0	CONCLUSIONS AND RECOMM	MENDATIONS	15
5.0	LIMITATIONS		18
6.0	REFERENCES		19
	LI	ST OF TABLES	
Tab		eristics, Nellie Grant Mint	11
Tab	Alternatives Nellie Gran	laterial Disposal/Disposition It Mine, Jefferson	16
	LIS	T OF FIGURES	
Figu		of the Nellie Grant Mine, Jefferson	. 2
Figu		rator Building and Shop Nellie Grant Montana	. 7
Figu		Storage Building Nellie Grant Mine,	. 8

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#### 1.0 INTRODUCTION

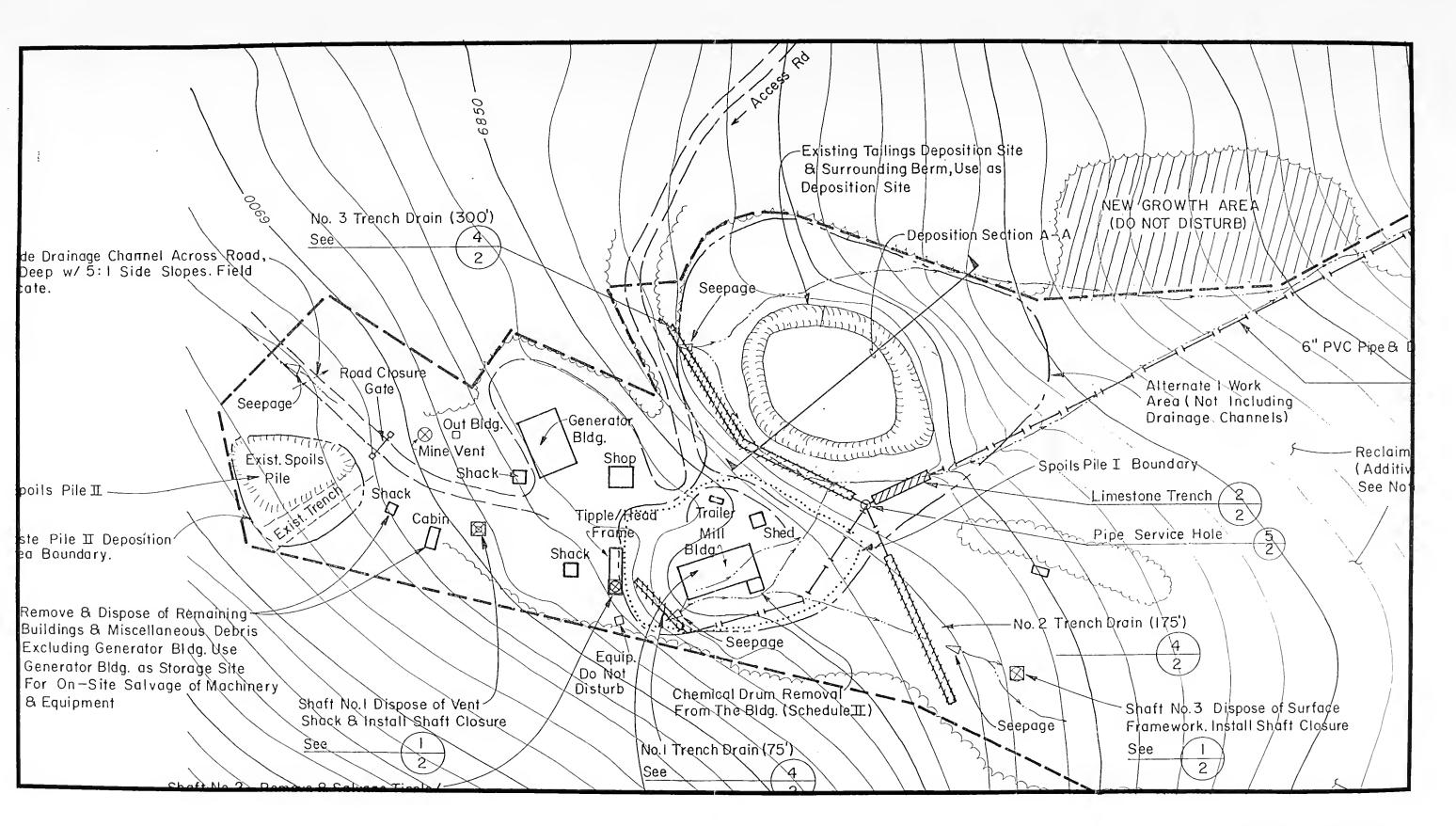
Chen-Northern, Inc. (Chen-Northern) prepared this report for the Montana Department of State Lands, Abandoned Mine Reclamation Bureau (AMR). This report presents the data and findings resulting from waste material sampling and analysis completed at the Nellie Grant Mine, Jefferson County, Montana. This work was performed to determine if these waste materials would pose a significant threat to human health and the environment during mine reclamation activities and to identify disposal/disposition alternatives for these materials.

The Nellie Grant Mine is located approximately 13 miles southwest of Helena, Montana. The mine was operated sporadically from about 1880 to about 1980. Minerals recovered at the mine were primarily lead and silver. Sparrow Resources, Ltd. was the last operator at the site and employed a crusher and mill to extract minerals from ore. The mill used chemical flotation agents to aid in mineral recovery.

During a preliminary assessment of the mine site, Robert Peccia and Associates (Peccia) (a contractor to AMR) discovered fifteen 55-gallon barrels stored in an small addition to the mill building (Figure 1). Because of the unknown contents of the barrels and other possible waste sources at the site, AMR contacted Chen-Northern to locate, sample and analyze waste materials at the site.

The analytical results from waste material sampling were intended to provide adequate data to characterize the waste materials as hazardous or non-hazardous and to possibly provide the identity of the waste materials. Once the waste material characteristics were classified, alternatives for material disposal/disposition could be identified prior to allowing mine reclamation activities to proceed.





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Chen-Northern's objectives in the waste material sampling and analysis investigation were to:

- Identify the location of liquid and solid waste materials stored in or near buildings at the site during a site reconnaissance;
- Prepare a sampling site safety plan and develop an analytical testing scheme for the waste materials;
- Collect representative samples for laboratory analysis from each identified waste;
- Characterize the wastes and determine if the wastes are classified as hazardous; and
- Develop disposal alternatives for each waste.

This report first describes the methods we used to investigate the waste materials at the site. This section is followed by a section describing the results of investigatory activities. The final section of the report contains our conclusions and recommendations for the ultimate disposition of waste material at the Nellie Grant Mine.

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#### 2.0 METHODS

This section describes our methods in completing site investigatory activities. The following subsections discuss our initial site reconnaissance, development of a site safety plan and testing scheme, waste sampling and analyses and generation of disposition alternatives for the wastes.

#### 2.1 SITE RECONNAISSANCE

A Chen-Northern project scientist completed a site reconnaissance on August 1, 1990. The purpose of the reconnaissance was to evaluate potential sampling hazards at the site and to identify the locations of waste material stored at the site. Our project scientist used site maps and site descriptions obtained from Peccia personnel to survey the site. In addition, a video tape was made during the reconnaissance and was subsequently shown to Chen-Northern laboratory and sampling personnel to inform them of potential hazards at the site.

Activities completed during the reconnaissance included the following:

- Inventorying the nature and approximate volume of wastes contained or spilled in or adjacent to the mill building addition, generator building and shop (Figure 1);
- ♦ Identifying the existence of product labels on all waste containers and recording the resultant information;
- ♦ Categorizing potential sampling safety hazards at the site; and
- Preparing maps of waste material storage areas.

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#### 2.2 SITE SAFETY PLAN AND ANALYTICAL TESTING SCHEME

Information obtained during the site reconnaissance was used to prepare a site safety plan for the imminent sampling event. In addition, product manufacturers and a former Sparrow Resources employee were contacted to determine the nature of potential waste material at the site (personal communication with Pat Bond, American Cyanamid and Mr. Kirk Miller, Chen-Northern, 1990). The resultant information was incorporated into the site safety plan. In addition, our safety plan was developed in general accordance with OSHA regulations (29 CFR 1910.120).

The site safety plan also included sections on waste material sampling protocol. Our sampling objective was to produce a set of samples representative of the waste materials and suitable for analysis. A copy of the site safety plan prepared for the sampling event is contained in Appendix A.

Ms. Kathy Smit, Division Manager of Chen-Northern's Billings, Montana analytical laboratory, designed the testing scheme for the waste material samples. Ms. Smit relied on waste material inventory information obtained during the site reconnaissance, product information from manufacturers and on our corporate experience in waste characterization. Analyses for each sample were designed to determine if the source waste material is classified as a hazardous waste pursuant to 40 CFR Part 261 et al. (EPA, 1990) and to assist in identifying the source materials.

#### 2.3 SAMPLING AND ANALYSIS

Chen-Northern field personnel completed sampling activities at the site on August 9 and 10, 1990. A total of 20 waste samples were obtained at locations depicted on Figures 2 and 3. Level C personal protective equipment and two vapor monitoring instruments were used by

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the sampling crew throughout the sampling event to ensure worker safety. A self contained breathing apparatus (SCBA) and a self rescuer were also available at the site for use in an emergency.

Prior to sampling the waste materials, the addition to the mill building was cordoned off with flagging to denote the sampling zone. A first aid station was established outside the sampling zone. Sampling procedures were completed in general accordance to the site safety plan (Appendix A) with the following exceptions:

- Waste container D (Figure 2) was not sampled because the container was apparently removed from the site sometime between the site reconnaissance visit and the sampling event; and
- Waste containers G (Figure 3) and sampling locations I and H (Figure 3) were added to the sample train to complete waste material characterization at the site.

#### 2.4 WASTE MATERIAL IDENTIFICATION AND DISPOSITION ALTERNATIVES

Analytical results from the 20 waste samples were evaluated with respect to the characteristics of a hazardous waste and to a variety of possible source products. The characteristics of a hazardous waste include toxicity, ignitability, reactivity and corrosivity. A list of possible source products for the waste materials was generated during site reconnaissance and site safety plan development activities. Possible source products at the site include:

- ♦ sodium sulfite anhydrous
- ♦ sodium isopropyl xanthate
- ♦ sodium silicate solution
- ♦ Aerofloat 340 Promoter (dithiophosphate family)
- ♦ Superfloc 330 (polyquaternary amine)
- ♦ methyl ethyl ketone (MEK)

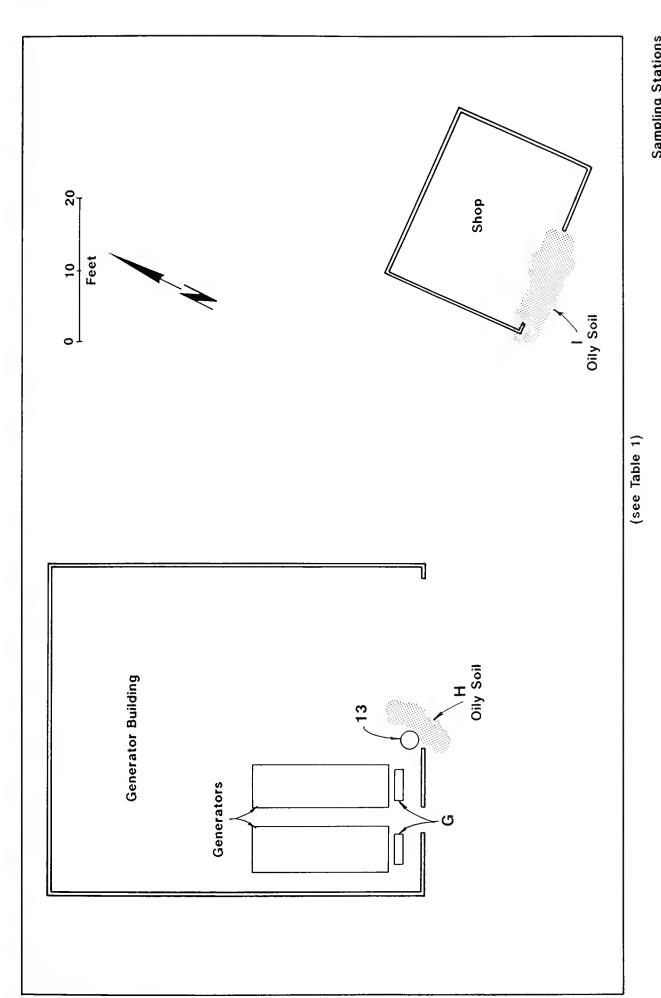
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Sampling Stations
Drum Storage Building
Nellie Grant Mine, Jefferson County, Montana
FIGURE 2

(see Table 1)

5 gallon Container

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Sampling Stations Generator Building and Shop Nellie Grant Mine, Jefferson County, Montana FIGURE 3

- ♦ methyl isobutyl carbinol (MIC)
- ♦ waste lubricating oil with or without polychlorinated biphenyl (PCB)

An evaluation of disposal/disposition alternatives for each of the 20 waste samples and their source material was generated using analytical data and State and Federal regulatory guidelines. In addition, persons at two active Montana mines, a waste oil processor, a hazardous waste shipping contractor and a chemical manufacturer were contacted to ascertain if their businesses would accept waste materials from the site (Appendix C).

#### 3.0 RESULTS AND DISCUSSION

This chapter presents a discussion of investigatory results obtained from waste material sampling at the Nellie Grant Mine. Included below are sections addressing the possible identification of waste materials and alternatives for disposal/disposition of the waste materials.

#### 3.1 WASTE MATERIAL IDENTIFICATION

Pertinent information about each waste material sample is presented on Table 1. This information was derived from physical observations and approximate volume measurements of the source waste material during sampling activities and from laboratory analytical results. Laboratory reports contained in Appendix B for each sample summarize the results of each analysis. The laboratory reports also list the EPA maximum concentrations for non-hazardous waste.

Table 1 indicates that there are at least five possible source "products" represented by the 20 waste material samples. These possible source products include:

- 1. Methyl Isobutyl Carbinol samples 1 and 11
- 2. Sodium Isopropyl Xanthate samples 5, 6, 7, 8, 9, and 10
- 3. Waste Oil samples 13 and G
- 4. Sodium Silicate Anhydrous samples A and F
- 5. Waste Oil in soil material samples H and I

Analytical results from samples 2, 3, 4, 12, B and C did not reveal a possible source product. In particular, the product labels on sample containers 4 and 12 did not match expected analytical results. Sample container 4 was labelled Aerofloat 340 Promoter which is dithiophosphate-based material (Appendix A). Analytical tests on sample 4 for total phosphorous indicated less that 4% phosphorous which apparently precludes sample 4 as

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Table 1 Waste Material Characteristics, Nellie Grant Mine Jefferson County, Montana

Hazardous Waste Characteristics									
Sample No. Location	Туре	Material Volume	тох	IGN	REA	COR	РСВ	Possible Material Identification	Disposal/Disposition Alternatives
1 mill building	L	≈ 10 gal		X			-	Methyl Isobutyl Carbinol (93%)	Reuse at operating mine or RCRA TSD
2 mill building	L	≈ 35 gal	X Lead			×	-	?	RCRA TSD
3 mill building	L	≈ 35 gal		Х	X	x	-	?	RCRA TSD
4 mill building	L	≈ 35 gal	X Lead					?	RCRA TSD
5 mill building	S	55 gal			х		_	Sodium Isopropyl Xanthate	Reuse at operating mine or disposal at RCRA TSD
6 mill building	S	55 gal			X			Sodium Isopropyl Xanthate	Reuse at Operating mine or disposal at RCRA TSD
7 mill building	S	55 gal			х		-	Sodium Isopropyl Xanthate	Reuse at Operating mine or disposal at RCRA TSD
8 mill building	S	55 gal			X		-	Sodium Isopropyl Xanthate	Reuse at Operating mine or disposal at RCRA TSD
9 mill building	S	17 gal			x		<b>-</b>	Sodium Isopropyl Xanthate	Reuse at Operating mine or disposal at RCRA TSD
10 mill building	S	55 gal			х		-	Sodium Isopropyl Xanthate	Reuse at Operating mine or disposal at RCRA TSD
11 mill building	L	50 gal		х			-	Methyl Isobutyl Carbinol (100%)	Reuse at operating mine or RCRA TSD
12 <sup>(1)</sup> mill building	L	≈ 30 gal					•	?	RCRA TSD
13 Generator Bld	L	≈ 20 gal					ND	Waste oil with Halogens & lead	Waste oil processor
A mill building	s	14 sacks					-	Sodium Sulfite Anhydrous	Reuse at operating mine or landfill
B(1) mill building	L	1/2 - 1 gal					-	?	RCRA TSD
C mill building	L	1/2 - 1 gal	X Lead			X	-	?	RCRA TSD
F mill building	S	≈ 0.1 yd <sup>3</sup>						Sodium Sulfite Anhydrous	Disposal at landfill
G Generator Bld	L	≈ 120 gal					ND	Waste oil with halogens	Waste oil processor
H Generator Bld	D	1-2 yd <sup>3</sup>		-	1		ND	Waste oil in soil	Landfarm
l shop	D	1-2 yd <sup>3</sup>	-			-	ND	Waste oil in soil	Landfarm

<sup>\*</sup>Key to Table 1 on following page

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#### KEY TO TABLE 1

Samples collected on August 9-10, 1990

Location: Refer to Figures 1, 2, and 3

Type: L = Liquid, S = Solid, D = Soil

Hazardous Waste Characteristics

TOX Toxicity - TCLP metals and Methyl Ethyl Ketone

IGN Ignitability - Flash point less than 140° F

REA Reactivity as Cyanide and/or Sulfide

COR Corrosivity pH less than 2.0 or more than 12.5

PCB Polychlorinated Biphenyls

RCRA TSD Facility permitted by the Resource Conservation and Recovery Act as a hazardous waste Treatment, Storage or Disposal facility.

Not analyzed

ND Not detected above detection limit

Refer to Appendix B for complete laboratory reports for each sample

(1) Laboratory holding time for volatile organics exceeded

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containing Aerofloat 340 Promoter (Appendix B). Sample container 12 was labelled Superfloc 330 which contains amines (nitrogen compounds) (Appendix A). The analytical test for total Kjeldahl nitrogen (Appendix B) indicated less than 5% nitrogen which apparently prevents sample 12 as being Superfloc 330.

Samples 13, G, H and I were preliminarily identified during the reconnaissance visit as consisting of waste oil. Samples 13 and G were liquid samples and were analyzed for several parameters including hazardous waste characteristics, MIC and PCBs (Appendix B). Samples G and 13 did not exhibit the characteristics of a hazardous waste but did exceed the 5 milligram per kilogram (mg/kg) allowable limit for lead in a reusable waste oil. PCBs were not detected in either sample 13 or G.

Samples H and I were apparently waste oil in a soil matrix. Total recoverable petroleum hydrocarbons in samples H and I were 64,000 mg/kg and 50,500 mg/kg, respectively. PCBs were not detected in either sample.

Of the 20 waste material samples, 12 exhibited the characteristics of a hazardous waste (Table 1). Three samples exhibited the toxicity characteristic. Lead was the only parameter that exceeded the toxicity regulatory level (Appendix B). In addition, three samples exhibited the characteristic of ignitability, seven samples were reactive and 3 samples were corrosive (Table 1). Although 12 samples tested positive as hazardous waste, it is possible that several sample sources could be considered product and be eligible for use in an active mining operation.

# 3.2 DISPOSAL/DISPOSITION ALTERNATIVES

The results of our investigation indicate there are viable options for both disposal and reuse of waste materials ate the site. These alternatives include:

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- Disposal at a sanitary landfill;
- ♦ Disposal at a RCRA-permitted treatment, storage or disposal (TSD) facility;
- Reuse at an active mining operation;
- Processing at a waste oil treatment facility; and
- ♦ Landfarming oily soil.

Table 1 presents alternatives for disposal/disposition for each sample source.

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#### 4.0 CONCLUSIONS AND RECOMMENDATIONS

It is our opinion that the 20 waste source materials identified during our investigation at the Nellie Grant Mine can be safely and legally accommodated prior to proceeding with mine reclamation activities. Ideally, waste materials identified at that could be considered "product" should be reused. The unidentified (unknown) waste materials at the site should be transported to a RCRA TSD facility.

Table 2 provides our specific recommendations for the ultimate disposition of the waste materials identified during our investigation. The following describes our rational for making these recommendations.

- Samples 1 and 11 are probably methyl isobutyl carbinol. Montana Tunnels Mine uses this chemical in their mill processes and is willing to accept the material (Appendix C).
- ♦ Samples 5, 6, 7, 8, 9 and 10 are probably sodium isopropyl xanthate. Montana Resources Inc., uses this material and will accept it for use in their mining operation. (Appendix C).
- ♦ Samples 13 and G are apparently waste oil. Montana Oil Processing will accept the oil for reprocessing and are available to transport the oil from the site to their facility (Appendix C).
- Samples A and F are probably sodium sulfite anhydrous. Because the material is non-hazardous, we feel it can be disposed of in a sanitary landfill. We recommend notifying a particular landfill of the material prior to disposal.

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TABLE 2

Recommended Waste Material Disposal/Disposition Alternatives
Nellie Grant Mine, Jefferson County, Montana

Possible Material Identification	Sample Numbers	Approximate Volume	Recommended Disposition
Methyl Isobutyl Carbinol	1 and 11	60 <u>+</u> gallons	Reuse at Montana Tunnels Mine
Sodium Isopropyl Xanthate	5, 6, 7, 8, 9 and 10	292 <u>+</u> gallons	Reuse at Montana Resources Inc. Mine
Waste Oil	13 and G	120 <u>+</u> gallons	Processing at Montana Oily Waste Processors
Anhydrous Sodium Sulfite	A and F	14 <u>+</u> sacks	Disposal at a Sanitary Landfill
Waste Oil in Soil	H and I	2 to 4 <u>+</u> cubic yards	Landfarm
Unknown	2, 3, 4, 12, B and C	137 <u>+</u> gallons	Disposal with Special Resources Management

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- ♦ Samples H and I are apparently waste oil in a soil matrix. We recommend excavating the oil-stained material at each sample location and landfarming the soil on-site to naturally degrade petroleum hydrocarbons.
- ♦ Samples 2, 3, 4, 12, B and C are unknown waste materials. Each of these samples except samples 12 and B exhibit the characteristics of a hazardous waste. Because our analytical tests did not provide positive identification of samples 12 and B and because samples 2, 3, 4 and C are classified as hazardous wastes, we recommend disposal at a RCRA TSD facility. Special Resources Management is available to arrange waste transport to a RCRA TSD facility.

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#### 5.0 LIMITATIONS

This work was performed in accordance with generally accepted practices of other consultants undertaking similar studies at the same time and in the same geographical area. Chen-Northern observed that degree of care and skill generally exercised by other consultants under similar circumstances and conditions. Chen-Northern's findings and conclusions must not be considered as scientific certainties, but as opinions based on our professional judgement concerning the significance of the data gathered during the course of the monitoring.

Prepared by:

Reviewed by:

K. Bill Clark

Hydrogeologist

David Hazen

Geologic Engineer

David Hazen

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#### 6.0 REFERENCES

American Cyanamid, 1990. Personal communication between Ms. Pat Bond of American Cyanamid and K. Bill Clark.

Miller, K., 1990. Personal communication between Mr. Miller, former Sparrow Resources employee and currently employed by Chen-Northern Inc., Boise, Idaho, and K. Bill Clark.

Sampling for Hazardous Materials, 1987. Environmental Protection Agency, Office of Emergency and Remedial Response, Hazardous Response Support Division, EPA Course 165.9; Course materials, Helena, Montana, August 18-20, 1987.

U.S. Environmental Protection Agency, 1990. Hazardous Waste Management System; Identification and Listing of Hazardous Waste; Toxicity Characteristics Revisions. Federal Register, Vol. 55, No. 61, March 29, 1990.

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# APPENDIX A SITE SAFETY PLAN

#### SITE SAFETY PLAN

# NELLIE GRANT MINE, JEFFERSON COUNTY, MONTANA

# DRUM/BARREL SAMPLING

#### A. GENERAL INFORMATION

Name of Facility: Nellie Grant Mine

Location: Approximately 13 miles southwest of Helena, Montana

CNI Project Manager: K. Bill Clark, Helena, Montana

Client: Montana Department of State Lands (DSL), Abandoned

Mine Bureau

Contact: Mr. Larry Marshall, 444-2074

Plan Prepared by:

Reconnaissance Visit:

K. Bill Clark - August 2, 1990

August 1, 1990 - K. Bill Clark

Background Information: DSL files located at Robert Peccia and Associates,

conversations with Kirk Miller, former Sparrow Resources employee -- overall incomplete site

background information

Proposed Sampling Date: August 8-10, 1990

Overall Hazard Summary: Moderate

#### B. OBJECTIVES OF THIS SITE SAFETY PLAN

The objective of this site safety plan is to protect Chen-Northern personnel from health and safety hazards they might experience in the performance of this task.

#### C. OBJECTIVES OF CHEN-NORTHERN PARTICIPATION

Chen-Northern's objectives are to:

- ♦ collect representative samples from liquid and solid wastes stored in thirteen 55-gallon drums and up to four 5-gallon containers stored at the site;
- submit the samples to an analytical laboratory for testing of hazardous waste characteristics and substance identification; and
- recommend disposal alternatives to DSL.

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#### D. SITE CHARACTERISTICS

Attachment A contains a site vicinity map, site map and a detail map of the drum storage area. Attachment B contains a table, referenced to the detail map, describing each waste container at the drum storage area.

A video tape was prepared during an August 1, 1990 site reconnaissance which shows the site layout and drum storage area. The wind direction during the morning of August 1 was from the east, upslope.

# 1. Site History

- The mine was first staked in the 1880's. Mining has occurred sporadically until about 1980. Minerals recovered were primarily lead and silver.
- Sparrow Resources, Ltd. was the last mining operation at the site. They used a portable crusher and mill to extract minerals. The mill was a flotation mill and used chemical floatation agents to recover minerals.
- According to Mr. Kirk Miller, the floatation agents included methyl ethyl ketone (MEK) or methyl ethyl carbinol, and xanthate compounds. Mr. Miller does not think that cyanide was used at the mill because gold concentrations were low to non-existent.

# 2. Site Description

- The project site consists of a mine entrance with a head frame, a tailings dump area, leach piles, and several buildings including cabins, a shop/office building, storage buildings, and a mill/crusher building (Attachment A).
- A 55-gallon steel drum is located at the front (south side) of the shop/office building (Attachments A and B). The top bung is open and the barrel is full of a dark liquid, presumably waste oil from two diesel generators. The ends of the barrel are bulged which probably indicates that some water is in the barrel.
- The drum storage area is located on the southeast side of the mill/crusher building (Attachment A). The drums are located in an approximately 15 foot by 15 foot addition to the main building. The addition is open to the east, has a dirt floor and has a door on the north wall open to the former mill.
- A road encircles the mill/crusher building (Attachment A). A relatively flat area is located on the north side of the building adjacent to the mine access road.

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- A surface drainage, discharging approximately one to five gallons per minute, flows along the south side of the mill/crusher building.
- Fifteen 55-gallon metal barrels (three empty with open tops), 14 sacks of a white chemical solid, and two 5-gallon containers (one appears empty) are stored in the addition to the mill/crusher building (Attachments A and B). Two 5-gallon plastic containers are located in front (east) of the addition; one upright and one on its side. A white powder spill is located on the dirt floor adjacent to the 14 chemical sacks.

#### 3. Utilities

• There is no evidence of "live" utilities at the site. Electric utility wires have been dismantled from power poles and cut. The two diesel engine units in the shop/office building probably supplied mine power needs.

# 4. Topography

The project site is located on an east-northeast facing slope at an elevation of about 6,850 feet above mean sea level. The area south and west of the mill/crusher building slopes up and the area east of the building slopes down (Attachment A).

# 5. Liquid and Solid Wastes Present

- Figures contained in Attachment A and the table contained in Attachment B should be reference for the following discussion.
- Although some of the barrels and containers are labelled, there is no assurance that the contents agree with the label.

#### a. Labelled Solid Wastes

- The 14 sacks on a shelf in the addition to the mill/crusher building, of which at least 8 are broken, are labeled *sodium sulfite anhydrous*. The white powder spilled adjacent to the sacks is possibly from these sacks, or former sacks.
- Six metal 55-gallon barrels (ID Nos. 5 through 10, Attachments A and B) are staged on the north side of the addition building. They have snap ring tops and are similar in appearance. This top configuration probably indicates that a solid material is stored in the barrels. At least one of the barrels is labelled sodium isopropyl xanthate. Barrel No. 10 has what appears to be two bullet holes in it. A white powder is visible through the holes.

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# b. Labelled Liquid Wastes

- The 5-gallon metal container No. E is labelled sodium silicate solution. The container is apparently empty.
- ♦ Barrel No. 4 is labelled Aerofloat 340 promoter, American Cyanamid. Barrel No. 3 is similar in appearance to barrel No. 4 (paint color scheme) and has a un-readable label caused by spillage and weathering.
- Barrel No. 12 has a partially readable label which apparently reads Superfloc 330, American Cyanamid. The top on this barrel is ripped open ≈25% and a dark liquid was observed. A hand-written date of 6-13-79 is located on the top portion of the barrel.

# c. Un-labelled Liquid Wastes

- Five-gallon plastic container No. B has an un-readable label and containers Nos. C and D are un-labelled. Mr. Miller recalls that either methyl ethyl ketone or methyl isobutyl carbinol used at the mine arrived in 5-gallon containers. It is may be possible that containers Nos. B, C and D contain these chemicals.
- Barrels Nos. 1 and 2 are un-labelled.
- ♦ Barrel No. 11 is un-labelled.
- Barrel No. 13, located at the shop/office building, is un-labelled but may possibly contain *waste oil* and water.

# d. Summary of Suspected Wastes at the Site

- ♦ sodium sulfite anhydrous
- sodium isopropyl xanthate
- sodium silicate solution
- ♦ Aerofloat 340 promoter
- ♦ Superfloc 330
- methyl ethyl ketone
- ♦ methyl isobutyl carbinol
- ♦ waste oil

#### E. HAZARD CHARACTERISTICS

Hazard characteristics for the suspected chemicals present at the Nellie Grant Mine are contained in Attachment C.

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#### F. RISK ASSESSMENT

The liquid and solid wastes to be sampled at the Nellie Grant Mine should present a moderate hazard to sampling personnel. This is based on the limited information obtained in the site reconnaissance and site review, and is based only on identified hazards.

The identified hazards to sampling personnel are inhalation of organic vapors and dusts, and eye and skin irritation. These hazards can be mitigated or controlled using personal protective equipment. This equipment will include full-face air purifying respirators (APR) and splash-protective suits and gloves. Sampling personnel will use a combination acid gas/organic vapor respirator cartridge. Non-sparking tools to open containers will be used to prevent the possibility of explosion. A fire extinguisher will be available at the Hot Line outside the drum storage building.

The expected wind direction is westerly (upslope) and into the open end of the drum storage building. Because the building will be well ventilated and because of the nature of the chemicals, the use of self contained breathing apparatuses (SCBA) may be unnecessary. However, an SCBA will be on-site and ready for use by the back-up sampler during all sampling activities.

#### G. SITE PROCEDURES

#### 1. Sampling Team Organization

PERSONNEL

ASSIGNMENT

K. Bill Clark Dave Hazen

#### 2. Site Entry Procedures

- Notify DSL and Robert Peccia and Associates personnel prior to the sampling event
- Cordon off the "Hot Zone" in front (east) of the drum storage building using marker tape
- Cordon off the "Decontamination Zone" and "Clean Zone"
- Don protective gear and assemble the decontamination pad in the order most practical for site conditions.
- ♦ Assemble all sampling equipment including air monitoring equipment at the entry point to the "Hot Zone"

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# 3. Safety Procedures

- ♦ Don protective gear in a manner to maximize skin protection from drips, spills and casual contacted with the liquid and solid wastes
- Protective gear should include from inside out: inner clothes, inner latex gloves, full face piece, hooded splash-resistant tyvex suit, safety boots, boot covers and outer chemical resistant gloves
- Have paper towels and/or absorbent pads available for wiping excess liquids from equipment and protective clothing
- Use plastic tarps/sheeting to cover barrel tops, sides and building floor
- The decontamination procedure should follow the process contained in Attachment D

# 4. Equipment List

plastic tarps
paper towel
absorbent pads
sample containers
plastic aprons
hooded tyvex suits
natural rubber gloves
rubber outer gloves
fire extinguisher
first aid kit
full face mask
drum thief
plastic bags
escape pack

tap water
decon tubs
brushes
sprayers
brass bung wrench
shovel
nitrile gloves
boot covers
eye wash station
air monitoring equipment
SCBA
plastic trowels
video camera

# 5. Air Monitoring

- Use air monitoring equipment to monitor atmosphere in drum storage building. If organic vapors are present in excess of 100 ppm, stop work, don SCBA and continue sample collection.
- Monitor vapor concentrations during collection of each sample.

organic vapor/acids gas filter cartridges

• If vapor concentrations exceed 500 ppm stop work until concentrations are reduced.

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# 6. Standard Operating Procedure

Survey atmosphere in building to determine if vapor/explosive hazard is present. Record data and determine if correct protective equipment is being used.

#### a. Liquids

- Place tarp over drum or container and cut hole over bung/spout. Open bung using the bung wrench. Place labelled sample container(s) on tarp.
- Insert drum thief and determine if solids are present. If not, reinsert thief, cover top and remove to determine how full drum is. Record this information.
- Reinsert thief into drum, cover top and remove. Allow excess liquid to drain off thief into drum, then begin to fill sample container(s). Repeat process until container(s) is full.
- Dispose of thief in drum and close bung/spout.
- Pick up tarp and proceed to decon pad. Dispose of tarp and outer gloves (if they can't be decontaminated) in a plastic garbage bag. Don new outer gloves.
- Repeat process for each drum/container.

#### b. Solids

- Open snap ring cover on drum. Place lid on plastic tarp on ground. Cover lid with plastic tarp and place labelled sample containers on tarp.
- Observe consistency of solid and fullness of container.
- Collect representative sample of solid with disposable trowel or stainless steel trowel/spoon.
- Pick up tarp and proceed to decon pad. Dispose of tarp and outer gloves in plastic garbage bag. Don new outer gloves.
- Repeat process for each drum/container.

#### 7. Decontamination

Discard plastic tarps, outer gloves and other contaminated protective equipment after sampling each container/drum.

# 8. Site Exit Procedures

Disassemble gear and markers.

# H. EMPLOYEE TRAINING REQUIREMENTS

At the time of job assignment, all employees assigned to the project shall receive a minimum of 40 hours of safety training.

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#### I. MEDICAL SURVEILLANCE

Medical surveillance for employees will be in accordance with the Chen-Northern written medical surveillance plan.

#### J. EMERGENCY INFORMATION

#### **Emergency Contacts**

- Dr. Maher phone 442-3570
- Poison Control Center phone 1-800-525-5042

### What to Report

State "This is an Emergency"
Your name and association
Telephone number to reach you
Your location
Name of person injured or exposed
Nature of emergency
Actions taken

# Route to Hospital

Route to St. Peter's Community Hospital marked on map in Attachment A

# First Aid/Safety Equipment

An eye wash station, first aid kit and fire extinguisher will be located at the drum location.

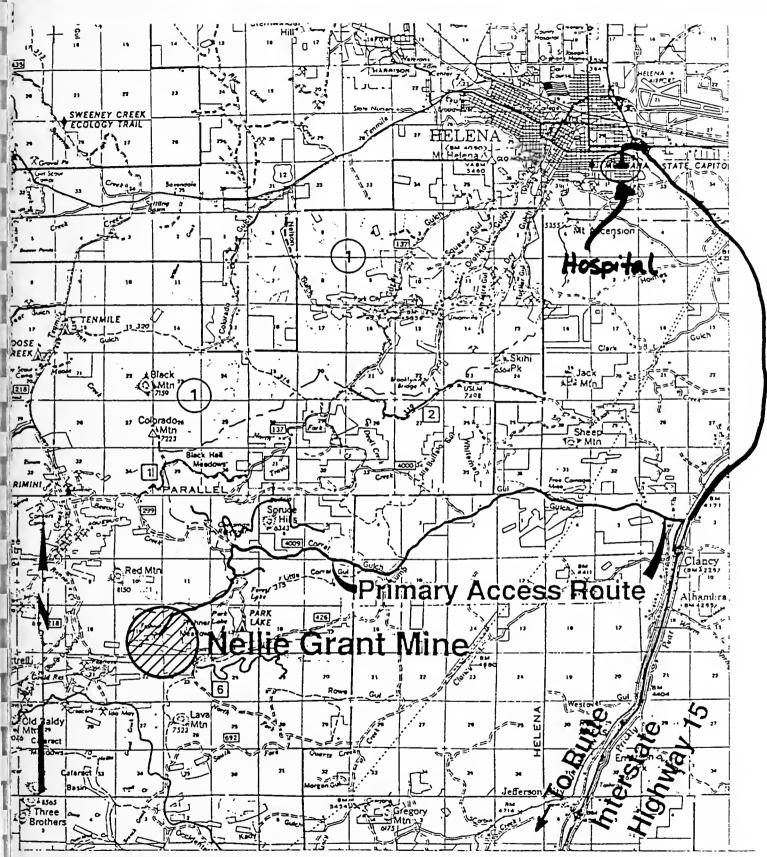
#### K. SAFETY PLAN APPROVAL

This safety plan was written for the use of Chen-Northern personnel performing sampling activities at the Nellie Grant Mine site, Jefferson County, Montana. This plan was written for specific site conditions, work tasks, dates and personnel specified and must be amended if conditions change.

PLAN REVIEWED BY:	DATE:	
PLAN REVIEWED BY:	DATE:	
PLAN APPROVED BY:		
CHEN-NORTHERN INDUSTRIA	L HYGIENIST	DATE:
CHEN-NORTHERN PROJECT M	ANAGER	DATE:
CHEN-NORTHERN DIVISION M	IANAGER	DATE:

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# ATTACHMENT A SITE MAPS



Nellie Grant Mine Reclamation Project
General Project Area Map
(from Helena National Forest Visitors Map)

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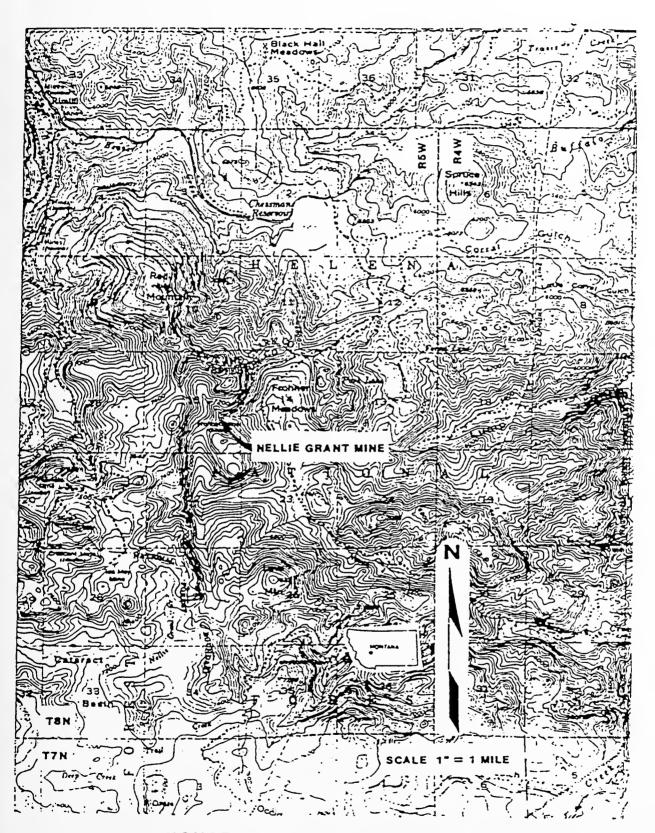


FIGURE 1 NELLIE GRANT MINE

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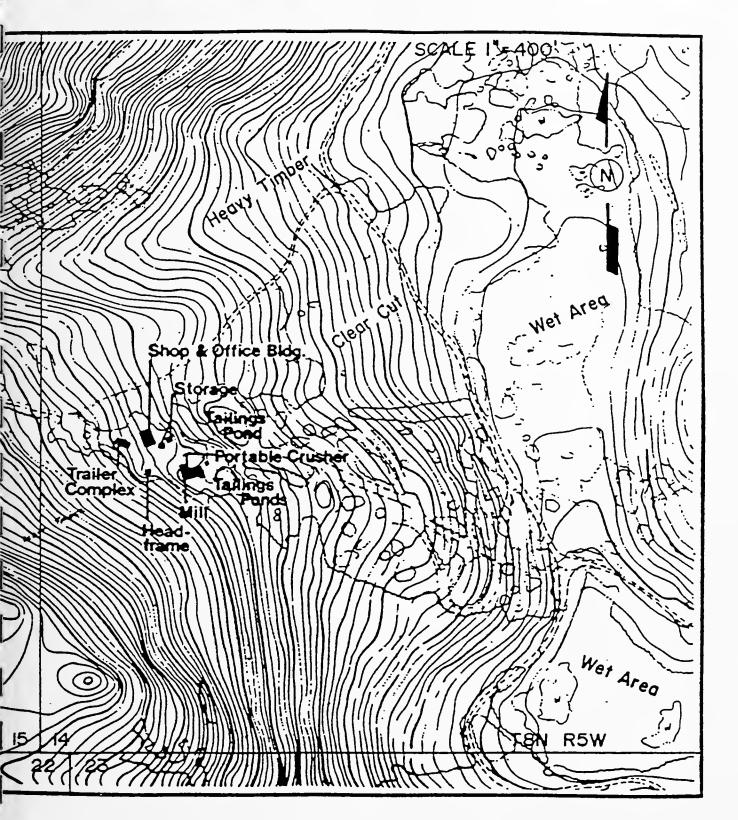


FIGURE 3 SURFACE FACILITIES

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## Chen Northern, Inc.

JOB. NO. 90-3133. N	JOB TITLE Nellie Grant Mine DATE 8-1-90 BY KW Clark
SUBJECT MAP OF DRUM	STORAGE AREA CHECKED SHEET OF
	headgate etc.
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CREATE IND	
CRIBHER BUILDING	ACCESS ROAD
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# ATTACHMENT B TABLE OF WASTE/CONTAINER INFORMATION

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### ATTACHMENT B

### WASTE/CONTAINER INFORMATION

Container Number	Container gallon/material	Тор Туре	Label	% Full	Remarks
Α	sacks	-	Sodium Sulfite Anhydrous	at least 8 sacks broken	14 total sacks
В	5/plastic	pour spout	? corrosive	5-10%	
С	5/plastic	pour spout	<u>-</u>	5%	no label
D	5/plastic	pour spout		50%	no label
Е	5/metal	cap	Sodium Silicate Solution	0-1% ?	appears empty
1	55/metal	bung		?	no label
2	55/metal	bung	-	?	no label
3	55/metal	bung	?Aerofloat	?	barrel color like #4
4	55/mctal	bung	Aerofloat 340 Promoter	?	American Cyanamid Label
5	55/metal	snap ring	Sodium Isopropyl Xanthate	?	
6	55/metal	snap ring	Sodium Isopropyl Xanthate	?	
7	55/metal	snap ring	Sodium Isopropyl Xanthate	?	
8	55/metal	snap ring	Sodium Isopropyl Xanthate	?	
9	55/metal	snap ring	Sodium Isopropyl Xanthate	?	
10	55/metal	snap ring	Sodium Isopropyl Xanthate	?	
11	55/metal	bung	_	100%	no label
12	55/metal	bung	Superfloc 330	50%	top open, date 6- 13-79
13	55/metal	bung	-	100%	barrel ends bulged

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## ATTACHMENT C HAZARD CHARACTERISTICS

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### ATTACHMENT C

#### HAZARD CHARACTERISTICS

Chemical	TLV	IDLH	LEL	ОТ	FP	First Symptom of over Exposure	Chronic Effects/ Other
Sodium Sulfite Anhydrous							addition of water could cause reaction
Sodium Isopropyl Xanthate	400		1.25			respiratory, eye and dermal irritation	depress CNS, headache, can decompose to carbon disulfide
Sodium Silicate Solution							
Aerofloat 340 Promoter					>200	burns eyes and skin	Use respiratory protection, splash gear, sulfur dioxide may form
Superfloc 330					>200		if heated will produce various oxides, slippery
Methyl Ethyl Ketone	200	3,000	2.0	10	21	irritation to eyes, nose and throat	CNS and respiratory damage, mint odor
Methyl Isobutyl Carbinol	25	2,000	1.0		106	irritation to eyes and skin, headache	eye and skin damage, mild odor

### <u>KEY</u>

TLV Threshold Limit Value, ppm, 8-hour time-weighted average IDLH Immediately Dangerous to Life and Health, ppm

Lower Explosive Limit, % Odor Threshold, ppm LEL OT FP Flash Point, °F

Central Nervous System CNS

Unknown



### TERIAL SAFETY DATA

Post-it" brand fax transmittal n	nemo 7671   # of pages > 7
To Jerry Bourser	From KAS
CO.CNI	co. CNI-Chen
Dept.	Phone #
Fax #	Fax #

PRO			
IDEN	JTIF	ICAT	ION

TRADE NAME: AERO® 343 Xanthate SYNONYMS: Sodium isopropyl xanthate CHEMICAL FAMILY: Alkyl xanthate salt MOLECULAR FORMULA: (CH3)2CHOC(S)SNa MOLECULAR WCT.:

158.2

WARNING

WARNING! HARMFUL IF ABSORBED THROUGH SKIN **DUST IRRITATING** CAUSES EYE AND SKIN IRRITATION

OSHA REGULATED COMPONENTS

COMPONENT	CAS. NO.	%	TWA/CEILING	REFERENCE
Sedium hydroxide	001010-73-2	1.5	2mg/M3 (celling)	OSHA/ACCIH
Isopropanol	000067-63-0	-0.5-1.0	400 ppm 500 ppm STEL	OSHA/ACCIH
Sodium suifide	001313-82-2	~1	not established	

NFPA HAZARD LATING

Fire

Health 2 1 Reactivity

Special

FIRE: Material that must be preheated

before ignition can occur.

HEALTH: Materials which on intense or continued exposure could cause temperary ineapacitation or possible residual injury unless prompt medical treatment

REACTIVITY: Materials which in themselves are normally stable, but which can become unstable at elevated temperatures and pressures or which may react with water with some release of energy

but not violently.

IEALTH HAZARD NFORMATION

EFFECTS OF OVEREXPOSURE:

The acute oral (rat) LD50 value for this material is calculated to be 569 mg/kg. The dermal (rabbit) LD50 value is between 400 and 1000 mg/kg. Skin or eye contact with solutions of this product may cause moderate eye and skin irritation. Airborne dust may cause significant eye, skin, or respiratory tract

Carbon disuifide may be released as a trace contaminant or as a decomposition product of xanthates. Overexposure to carbon disulfide may produce eye, skin and respiratory tract irritation, skin sensitization, dizziness, headache, degeneration of peripheral nerves, manic depressive psychosis and cardiovascular disorders.

Taxicalogy information on regulated components of this product is as follows:

MERGENCY PHONE: 201/835-3100

MERICAN CYANAMID COMPANY, 1 CYANAMID PLAZA, WAYNE, NEW JERSEY 07470

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AERO® 343 Xanthate

MSDS NO. 0293-05

PAGE 2 OF 4

Acute overexposure to sodium hydroxide mists or dusts causes severe respiratory imitation. A solution of sodium hydroxide can produce irreversible damage to eyes and skin.

Acute overexposure to isopropanol vapor may cause some irritation of the eyes and respiratory system. Repeated or prolonged overexposure to bopropanol vapor may cause central nervous system depression, resulting in headache, dizziness, nausea and staggered galt. The liquid isopropanol is a moderate eye irritant. The oral (rat) and dermal (rabbit) LD50 values for isopropanol are 4.7-5.8 g/kg and 12.8 g/kg, respectively. The LC50 after a four hour exposure to rats is greater than 12,000 ppm. Sodium sulfide irritates skin and mucous membranes. This material

liberates hydrogen sulfide upon contact with acids.

in case of skin contact, remove contaminated clothing without delay. flush skin thoroughly with water. Do not reuse clothing without

In case of eye contact, immediately imigate with plenty of water for 15 minutes. Obtain medical attention if initation persists.

#### **XPOSURE** ONTROL METHODS

RST AID:

Where this material is not used in a closed system, good enclosure and local exhaust ventilation should be provided to control exposure. Food, beverages, and tobacco products should not be carned, stored, or consumed where the material is in use. Denote eating, drinking, or smoking, wash face and hands with soap and water. Avoid skin contact. Protective clothing such as impervious gloves, apron, workpants, long sleeve work shirt, or disposable coveralls are recommended to prevent skin contact. For operations where eye or face contact can occur, wear eye protection such as chemical splash proof goggles or face shield. Eyewash equipment and safety shower should be provided in areas of potential exposure. Where exposures are below the Permissible Exposure Limit (PEL), no respiratory protection is required. Where exposures exceed the PEL, use respirator approved by NIOSH for the material and level of exposure. See "CUIDETO INDUSTRIAL RESPIRATORY PROTECTION" (NIOSH).

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LEROP 343 Xanthate

MSDS NO. 0293-05 PAGE 3 OF 4

REAND	FLASH POINT:	Not Applicable			
PLOSION AZARD FORMATION	FLAMMABLE LIMITS (% BY VOL):	Lower - 1.25; Upper - 50.0 (values for carbon disulfide)			
	AUTOIGNITION TEMP.	248 F (120 C) (value for carbon disulfide)			
	DECOMPOSITION TEMP.	428-464 F(220-240 C)			
,	FIRE FIGHTING:	Use carbon dioxide, dry chemical or large quantities of water to extinguish fires. Heat causes decomposition to vapor of carbon disulfide. Wear self-contained, positive pressure breathing apparatus and full firefighting protective clothing. Solid xanthates are stable when kept cool and dry. However, exposure to heat and moisture can cause decomposition to flammelie and explosive vapor of carbon disulfide. Since xanthates decompose in sulution, even at room temperature, fire and explosion hazards can develop with aging. The moisture precautions do not apply to the product when diluted according to the Cyanamid Product Bulletin.			
ACTIVITY DATA	STABILITY: CONDITIONS TO AVOID:	Unstable Exposure of the solid xanthate to heat or moisture and heating or aging of xanthate solutions.			
	POLYMERIZATION: CONDITIONS TO AVOID:	Will Not Occur None known			
	INCOMPATIBLE MATERIALS:	Strong acids, oxidizing agents, moisture.			
	HAZARDOUS DECOMPOSITION PRODUCTS:	Heat or moisture will liberate carbon disulfide. Therrial decomposition may produce sarbon monavide carbon disulfide.			
YSICAL OPERTIES	APPEARANCE AND ODOR:	Yellow pellets or powder; alight disagreeable ador			
	BOILING POINT:	Not Applicable			
	MELTING POINT:	451-462 F(233-239 C)			
	VAPOR PRESSURE:	Not Applicable			
	SPECIFIC GRAVITY:	Not Available			
	VAPOR DENSITY:	Not Applicable			
	% VOLATILE (BY VOL):	~1.5			
	OCTANOL/H2O PARTITION COEF.:	Not Applicable			
	pH:	Not Applicable			
	SATURATION IN AIR (BY VOL):	Not Applicable			
	EVAPORATION RATE:	Not Applicable			
	SOLUBILITY IN WATER	Appreciable			

Marvin A. Friedman, Ph.D. Director of Toolcology and Product Safety

information is given without any warranty or representation. We do not assume any legal responsibility for same, to we give permission, inducement, or recommendation to practice any patented invention without a license, if ered solely for your consideration, investigation and verification. Before using any product read its label.

### PAGE 1 OF 4



CYANAMID

MSDS NO. 1922-04 DATE: 05/04/89.

PRODUCT	TRADE NAME:	S	UPERFLOC	330	Flocculant	
PENTIFICATION	SYNONYMS:	Po	olyquatemary a	mine ln	water solution	
	CHEMICAL FAMILY:	С	ationic polyme			
1	MOLECULAR FORM	IULA: M	lixture			
	MOLECULAR WCT.	: M	lbdure			
WARNING	IMPORTANTI SPIL	LS OF TH	HIS PRODUC	ARE V	ERY SLIPPERY.	
OSHA REGULATED	COMPONENT	CAS	. NO.	%	TWA/CEILING	REFERENCE
PMPONENTS	No Permissible Exposure Limits (PEL/TLV) have been established by OSHA or ACGIH.					
NFPA HAZARD	Fire 1 Health 0 0 Res Special	activity	before Ignition HEALTH: Mate would offer no ordinary comb REACTIVITY: I stable, even u	can occerials who hazard oustible in Materials ander fire	nich on exposure under beyond that of	
JAITH HAZARD FORMATION	EFFECTS OF OVEREXPOSURE:	The acute oral (rat) LD50 for the active ingredient in this product is 5.0 ml/kg. Subchronic feeding studies in rats and dogs did not produce any significant adverse effects when the active ingredient of this product was evaluated at a dietary concentration of 1 or 4 percent.  No specific first aid procedures are necessary for accidental exposu to this product.				its and then the
	FIRST AID:					iental exposure
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MERGENCY PHONE: 201/835-3100

AMERICAN CYANAMID COMPANY, 1 CYANAMID PLAZA, WAYNE, NEW JERSEY 07470

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ואוטעט וזער ו זעגיעק

INULAUIT

PERFLOC® 330 Flocculant

POSURE NTROL METHODS Engineering controls are not usually necessary if good hygiene practices are followed. Before eating, drinking, or smoking, wash face and hands thoroughly with soap and water. Avoid unnecessary skin contact. Impervious gloves are recommended to prevent prolonged skin contact. For operations where eye or face contact can occur, eye protection is recommended.

REACTIVITY DATA	FLAMMABLE LIMITS (% BY VOL):  AUTOIGNITION TEMP:  DECOMPOSITION TEMP:  FIRE FIGHTING:  STABILITY: CONDITIONS TO AVOID:  POLYMERIZATION: CONDITIONS TO AVOID:  INCOMPATIBLE MATERIALS:  HAZARDOUS  DECOMPOSITION:	Will Not Occur None known Strong oxidizing agents: this product corrodes iron, copper and aluminum.
REACTIVITY DATA	DECOMPOSITION TEMP: FIRE FIGHTING:  STABILITY: CONDITIONS TO AVOID: POLYMERIZATION: CONDITIONS TO AVOID: INCOMPATIBLE MATERIALS: HAZARDOUS	Not Available  Use water spray, carbon dioxide or dry chemical to extinguish fires Use water to keep containers cool. Wear self-contained, positive pressure breathing apparatus.  Stable None known  Will Not Occur None known  Strong oxidizing agents: this product corrodes iron, copper and aluminum.
REACTIVITY DATA	STABILITY: CONDITIONS TO AVOID: POLYMERIZATION: CONDITIONS TO AVOID: INCOMPATIBLE MATERIALS: HAZARDOUS	Use water spray, carbon dioxide or dry chemical to extinguish fires Use water to keep containers cool. Wear self-contained, positive pressure breathing apparatus.  Stable None known  Will Not Occur None known  Strong oxidizing agents: this product corrodes iron, copper and aluminum.
EACTIVITY DATA	STABILITY: CONDITIONS TO AVOID: POLYMERIZATION: CONDITIONS TO AVOID: INCOMPATIBLE MATERIALS: HAZARDOUS	Use water to keep containers cool. Wear self-contained, positive pressure breathing apparatus.  Stable None known  Will Not Occur None known  Strong oxidizing agents: this product corrodes iron, copper and aluminum.
	POLYMERIZATION: CONDITIONS TO AVOID: INCOMPATIBLE MATERIALS: HAZARDOUS	None known  Will Not Occur None known  Strong oxidizing agents: this product corrodes iron, copper and aluminum.
	INCOMPATIBLE MATERIALS: HAZARDOUS	None known  Strong oxidizing agents: this product corrodes iron, copper and aluminum.
	MATERIALS: HAZARDOUS	aluminum.
	DECOMPOSITION PRODUCTS:	Thermal decomposition or combustion may produce carbon monoxide, carbon dioxide, ammonia, oxides of nitrogen and/or hydrogen chloride.
PHYSICAL PROPERTIES	APPEARANCE AND ODOR:	Amber liquid; slight, amine odor
	BOILING POINT:	Not Available
	MELTING POINT:	Freezing point: 0 F; - 18C
	VAPOR PRESSURE:	Not Available
	SPECIFIC GRAVITY:	1.08-1.18
	VAPOR DENSITY:	Not Available
•	% VOLATILE (BY VOL):	~50 (water by weight)
	OCTANOL/H2O PARTITION COEF.:	Not Available
	pH:	5 - 7
	SATURATION IN AIR (BY VOL):	Not Available
	EVAPORATION RATE: SOLUBILITY IN WATER:	Not Available  Complete

UPERFLOC® 33	0 Flocculant	INDUSTRUCTURE TRUCTURE
SPILL OR LEAK PROCEDURES	CASE MATERIAL IS RELEASED OR SPILLED:	Spills of this product are very slippery. Spilled material should be absorbed onto an inert material and scooped up. The area should be thoroughly flushed with water and scrubbed to remove residue. If slipperiness remains apply more dry-sweeping compound.
ASTE DISPOSAL	Disposal must be made in acregulations.	accordance with applicable governmental
ECIAL LECAUTIONS	HANDLING AND NO STORAGE/OTHER:	ONE
D.T. SHIPPING	PROPER SHIPPING NAME:	NOT APPLICABLE/NOT REGULATED
	HAZARD CLASS:	NOT APPLICABLE
	UN/NA:	NOT APPLICABLE
-d)	D.O.T. HAZARDOUS SUBSTANCES:	(Reportable Quantity of Product) NOT APPLICABLE
1	D.O.T. LABEL REQUIRED:	D: NOT APPLICABLE
SCA FORMATION	This product is manufactured Toxic Substances Control Ac	ed in compliance with all provisions of the act, 15 U.S.C.
NVIRONMENTAL NFORMATION	The following components a Section 313 of Title III and of	are defined as toxic chemicals subject to reporting requirements of of 40 CFR 372 or subject to other EPA regulations.
-11	COMPONENT CAS. NO	SARA TITLE III  10. % TPQ (Ibs.) RQ (Ibs.) S313 RCRA TSCA 128
	This product does not contain any components regulated under these sections of the EPA	
71	PRODUCT CLASSIFICATI	TION UNDER SECTION 311 OF SARA
	Not Applicable under SA	ARA TITLE III
11		
•)		•
]]		
11		Marvin A. Friedman, Ph.D., Director of Toxicology and Product Safety
ndo we give permission,	Inducement, or recommendation to pra	. We do not assume any legal responsibility for same, ractice any patented invention without a license. ration. Before using any product read its label.

PAGE 1 OF 3



## MATERIAL SAFETY DATA

MSDS NO. 0413-03 DATE: 03/08/89

		_	477 Aqueous	
SYNONYMS:	None			
CHEMICAL FAMILY	: Dithlophosph	nate		
MOLECULAR FORM	AULA: Mixture			
MOLECULAR WGT.	.: Mixture			
DANGERI CAUSI	es burns of eyes	AND SKIN		
COMPONENT	CAS. NO.	%	TWA/CEILING	REFERENCE
Sodium hydroxlde	001310-73-2	~ .5	2mg/M3 (ceiling)	OSHA/ACCIH
Fire 1 Health 3 0 Re Special	before Igni HEALTH: I eactivity serious ter prompt ma REACTIVIT stable, eve	Ition can occ Materials wh nporary or redical treatn Y: Materials en under fire	cur.  Jich on short exposure of esidual injury even thousant were given.  Which in themselves a exposure conditions,	ugh
EFFECTS OF OVEREXPOSURE:	for a similar product Material was corrost Direct contact with Irritation. Contact with acid in sulfide has a strong to smell the gas and Therefore odor is art to hydrogen sulfide Irritation, rapid develents of hydrogens gait, neurological da Acute overexposure respiratory irritation	t are 3.54 g/ive (skin) in this material nay cause lib rotten-egg of exposure we unreliable it gas may cause opment of sulfide may cause and get o sodium it. A solution	kg and 7.07 g/kg, resprabblts.  I may cause severe eyesteration of hydrogen subdor, however, some positional deaden the sense of indicator of exposure. It is essevere eye or respir coma and respiratory frause headache, dizzingastritis.  The hydroxide mists or dust of sodium hydroxide care	ectively.  and skin  lifide. Hydrogen eople are unable ismell.  Overexposure ratory tract allure. Low ess, staggering a causes severe
FIRST AID:	In case of skin conta Wear impervious gle Do not omit cleanin not reuse clothing w leatherware. In case of eye conta	act, remove oves. Cleans g hair or und without laund act, Immedia	contaminated clothing se skin thoroughly with der fingemails if contain lering. Do not reuse cottely irrigate with plenty	soap and water. ninated. Do ontaminated
	CHEMICAL FAMILY MOLECULAR FORA MOLECULAR WGT.  DANGERI CAUSI COMPONENT Sodium hydroxide  Fire 1 Health 3 0 Re Special  EFFECTS OF OVEREXPOSURE:	CHEMICAL FAMILY: Dithlophosph  MOLECULAR FORMULA: Mixture  MOLECULAR WGT.: Mixture  DANGERI CAUSES BURNS OF EYES  COMPONENT CAS. NO.  Sodium hydroxide 001310-73-2  Fire before Ign 1 HEALTH: I Health 3 0 Reactivity serious terprompt me Special REACTIVIT stable, ever and which  EFFECTS OF Acute oral (rat) and for a similar product Material was corrost Direct contact with Irritation.  Contact with acid me sulfide has a strong to smell the gas and Therefore odor is an to hydrogen sulfide irritation, rapid dever levels of hydrogen significant production in the production of the producti	CHEMICAL FAMILY: Dithlophosphate  MOLECULAR FORMULA: Mixture  MOLECULAR WGT.: Mixture  DANGERI CAUSES BURNS OF EYES AND SKIN  COMPONENT CAS. NO. %  Sodium hydroxide 001310-73-25  FIRE: Material that murbefore Ignition can occur and the molecular management of the prompt medical treatment of the prompt medical tre	CHEMICAL FAMILY: Dithlophosphate  MOLECULAR FORMULA: Mixture  MOLECULAR WGT.: Mixture  DANGER! CAUSES BURNS OF EYES AND SKIN  COMPONENT CAS. NO. % TWA/CEILING  Sodium hydroxide 001310-73-25 2mg/M3 (ceiling)  Fire before ignition can occur.  HEALTH: Materials which on short exposure serious temporary or residual injury even tho prompt medical treatment were given.  Special REACTIVITY: Materials which in themselves a stable, even under fire exposure conditions, and which are not reactive with water.  EFFECTS OF OVEREXPOSURE: Acute oral (rat) and acute dermal (rabbit) LD50 values for a similar product are 3.54 g/kg and 7.07 g/kg, resp. Material was corrosive (skin) in rabbits.  Direct contact with acid may cause liberation of hydrogen su sulfide has a strong rotten-egg odor, however, some pot on smell the gas and exposure will deaden the sense of Therefore odor is an unreliable indicator of exposure. To hydrogen sulfide gas may cause severe eye or respir irritation, rapid development of coma and respiratory flevels of hydrogen sulfide may cause headache, dizzin gait, neurological damage and gastritis.  Acute overexposure to sodium hydroxide mists or dust respiratory irritation. A solution of sodium hydroxide cireversible damage to eyes and skin.  FIRST AID: In case of skin contact, remove contaminated clothing Wear impervious gloves. Cleanse skin thoroughly with Do not omit cleaning hair or under fingernalis if contamor reuse clothing without laundering. Do not reuse contaminated clothing more contaminated clothing more contaminated clothing wear impervious gloves. Cleanse skin thoroughly with Do not omit cleaning hair or under fingernalis if contamor reuse clothing without laundering. Do not reuse clothing without laundering. Do not reuse contaminated clothing more contaminated clothing more contaminated clothing more contaminated clothing without laundering. Do not reuse clothing without laundering.

CONTROL METHODS

Utilize a closed system process where feasible. Where this material

**EMERGENCY PHONE: 201/835-3100** 

AMERICAN CYANAMID COMPANY, 1 CYANAMID PLAZA, WAYNE, NEW JERSEY 07470

AERO® Promoter 3477 Aqueous

MSDS NO. 0413-03

PAGE 2 OF 3

is not used in a closed system, good enclosure and local exhaust ventilation should be provided to control exposure. Food, beverages, and tobacco products should not be carried, stored, or consumed where this material is in use. Before eating, drinking, or smoking, wash face and hands with soap and water. Prevent eye and skin contact. For operations where eye or skin contact can occur wear the special protective equipment specified below. Prevent contamination of skin or clothing when removing protective equipment. Provide eyewash fountain and safety shower in close proximity to points of potential exposure. For operations where Inhalation exposure can occur, a NIOSH approved respirator recommended by an industrial hygienist may be necessary. A full facepiece respirator also provides eye and face protection. Special protective equipment - To prevent skin contact wear skin protection, such as impervious gloves, apron, workpants, long sleeve workshirt, or disposable coveralls. To prevent eye contact wear eye protection such as chemical splash proof goggles. or face shield.

FIRE AND EXPLOSION	FLASH POINT: METHOD:	>200 f (>93.3 C) Setaflash Closed Cup
JAZARD IFORMATION	FLAMMABLE LIMITS (% BY VOL):	Not Available
	AUTOIGNITION TEMP:	Not Available
	DECOMPOSITION TEMP:	Not Available
	FIRE FIGHTING:	Use water, carbon dioxide or dry chemical to extinguish fires. Wear self-contained, positive-pressure breathing apparatus and full firefighting protective clothing. See Exposure Control Methods for special protective clothing. Sulfur dioxide or oxides of phosphorus may be formed under fire conditions. Do not flush to sewer which may contain acid. This could result in generation of toxic and explosive hydrogen sulfide gas.
ACTIVITY DATA	STABILITY: CONDITIONS TO AVOID:	Stable None known
	POLYMERIZATION: CONDITIONS TO AVOID:	Will Not Occur None known
1	INCOMPATIBLE MATERIALS:	This product contains a neutralized dithioacid. Avoid contact with strong oxidizing agents and mineral acids.
]	HAZARDOUS DECOMPOSITION PRODUCTS:	Thermal decomposition or combustion may produce carbon monoxide, earbon dioxide, hydrogen sulfide and/or oxides of sulfur and phosphorus.
HYSICAL ROPERTIES	APPEARANCE AND ODOR:	Clear, pale yellow-dark amber, mobile liquid; noticeable odor
	BOILING POINT:	Not Available
	MELTING POINT:	<-5 F(<-20.6 C)
_	VAPOR PRESSURE:	Similar to water
1	SPECIFIC GRAVITY:	1.12 @ 25 C
	VAPOR DENSITY:	Similar to water
	% VOLATILE (BY VOL):	50 (water by weight)
]	OCTANOL/H2O PARTITION COEF.:	Not Available

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## ATTACHMENT D DECONTAMINATION PROCEDURE

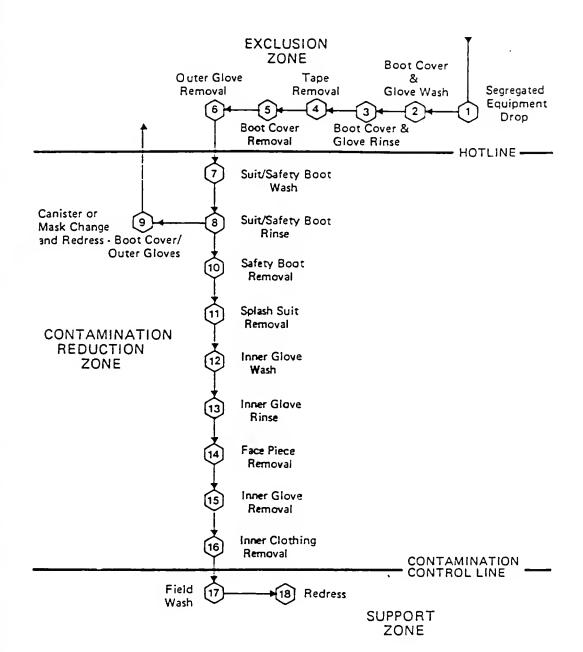
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S.O.P. No. 7

ROCESS DECON PROCEDURES

#### MAXIMUM DECONTAMINATION LAYOUT

#### LEVEL C PROTECTION



Source: October 85 NIOSH/OSHA/USCG/EPA-Occupational Sufety of Health Guidance Manual for-Hazardus unsk Site Activities



### FSOP 7: MAXIMUM MEASURES FOR LEVEL C DECONTAMINATION

Station 1:	Segrated Equipment Drop	1.	Deposit equipment used on site (tools, sampling devices and containers, monitoring instruments, radios, clipboards, etc.) on plastic drop cloths or in different containers with plastic liners. Segregation at the drop reduces the probability of cross contamination. During hot weather operations, a cool down station may be set up within this area.
Station 2:	Boot Cover and Glove Wash	2.	Scrub outer boot covers and gloves with decon solution or detergent and water.
Station 3:	Boot Cover and Glove Rinse	3.	Rinse off decon solution from station 2 using copious amounts of water.
Station 4:	Tape Removal	4.	Remove tape around boots and gloves and deposit in container with plastic liner.
Station 5:	Boot Cover . Removal	5.	Remove boot covers and deposit in containers with plastic liner.
Station 6:	Outer Glove Removal	6.	Remove outer gloves and deposit in container with plastic liner.
Station 7:	Suit and Boot Wash	7.	Wash splash suit, gloves, and safety boots. Scrub with long-handle scrub brush and decon solution.
Station 8:	Suit and Boot, and Glove Rinse	8.	Rinse off decon solution using water. Repeat as many times as necessary.
Station 9:	Canister or Mask Change	9.	If worker leaves exclusion zone to change canister (or mask), this is the last step in the decontamination procedure. Worker's canister is exchanged, new outer gloves and boot covers donned, and joints taped worker returns to duty.
Station 10:	Safety Boot Removal	10.	Remove safety boots and deposit in container with plastic liner.
Station 11:	Splash Suit Removal	11.	With assistance of helper, remove splash suit. Deposit in container with plastic liner.
Station 12:	Inner Glove Rinse	12.	Wash inner gloves with decon solution.
Station 13:	Inner Glove Wash	13.	Rinse inner gloves with water.
Station 14:	Face Piece Removal	14.	Remove face piece. Deposit in container with plastic liner. Avoid touching face with fingers.
Station 15:	Inner Glove Removal	15.	Remove inner gloves and deposit in lined container.

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#### FSOP 7: MAXIMUM MEASURES FOR LEYEL C DECONTAMINATION

Station 16: Inner Clothing Removal

16. Remove clothing soaked with perspiration and place in lined container. Do not wear inner clothing off-site since there is a possibility that small amounts of contaminants might have been transferred in removing the fullyencapsulating suit.

Station 17: Field Wash

17. Shower if highly toxic, skin-corrosive or skinabsorbable materials are known or suspected to be present. Wash hands and face if shower is not available.

Station 18: Redress

18. Put on clean clothes.

#### FSOP 7: MINIMUM MEASURES FOR LEVEL C DECONTAMINATION

Station 1: Equipment Drop

 Deposit equipment used on-site (tools, sampling devices and containers, monitoring instruments, radios, clipboards, etc.) on plastic drop cloths. Segregation at the drop reduces the probability of cross contamination. During hot weather operations, a cool down station may be set up within this area.

Station 2: Outer Garment, Boots, and Gloves Wash and Rinse Scrub outer boots, outer gloves and splash suit with decon solution or detergent water. Rinse off using copious amounts of water.

Station 3: Outer Boot and Glove Removal

Remove outer boots and gloves. Deposit in container with plastic liner.

Station 4: Canister or Mask Change

4. If worker leaves exclusive zone to change canister (or mask), this is the last step in the decontamination procedure. Worker's canister is exchanged, new outer gloves and boot covers donned, joints taped, and worker returns to duty.

Station 5: Boot, Gloves and Outer Garment Removal Boots, chemical-resistant splash suit, inner gloves removed and deposited in separate containers lined with plastic.

Station 6: Face Piece Removal

Facepiece is removed. Avoid touching face with fingers, 'Facepiece deposited on plastic sheet.

Station 7: Field Wash

Hands and face are thoroughly washed. Shower as soon as possible.

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# APPENDIX B LABORATORY REPORTS

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## hen-Northern, Inc.

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600 SOUTH 25TH STREET P. O. BOX 30615 BILLINGS, MT 59107 (406) 248-9161 FAX (406) 248-9282

## TECHNICAL REPORT



REPORT TO:

CHEN-NORTHERN, INC. ATTN: MR. BILL CLARK

P 0 BOX 4699

HELENA, MT 59601

\*Revised: 10/18/90

**DATE:** October 2, 1990

JOB NUMBER: 87-911 SHEET: 1 OF 3

INVOICE NO.: 104215

REPORT OF:

Solid Waste Analysis - Nellie Grant Mine

#### Sample Identification:

On August 14, 1990, these samples of solid waste (laboratory numbers 106590 and 106591) were delivered to our laboratory for analysis. The field personnel identified these samples as possibly waste oil. The tests were conducted in accordance with the U.S. Environmental Protection Agency Manual SW-846, <u>Test Methods for Evaluating Solid Waste</u>, 3rd Edition, November 1986 and 40 CFR Part 302 Method 1311 published June 29, 1990. The results of the analysis are shown on the following pages.

A < sign indicates less than the reported value was present in the sample.

\*pH value for Lab No. 106591 was changed. Also, the comments changed at the bottom of page 3.

Reviewed by

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October 2, 1990 Job No. 87-911 Sheet 2 of 3

Lab No.: Sample Description: Date and Time Sampled: Collected by:

106590 Barrel 13 8/09/90 1525 Bill Clark

TOXICITY	Total Metals, mg/kg	Measured TCLP, mq/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	<2 <20 <1 7 143 <0.17 12 <4	<0.035 0.7 <0.005 0.04 0.94 0.0008 <0.035 <0.02	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
Methyl Ethyl Ketone	<200		200
IGNITABILITY			
Closed Cup Flash Point	>150 ° F		140 ° F
REACTIVITY, mg/kg			
Reactive Cyanide as CN Reactive Sulfide as S	<0.10 13		250 500
CORROSIVITY, standard units	3		
рН	6.8		2.0-12.5
Polychlorinated Biphenyl Total Halogens as Cl Methyl Isobutyl Carbinol	<50 mg/kg 80 mg/kg <400 mg/kg		

This sample does not exhibit the characteristics of a hazardous waste. However, it does exceed the allowable limits for lead in a waste oil.

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October 2, 1990 Job No. 87-911 Sheet 3 of 3

Lab No.:

Sample Description:

Date and Time Sampled: Collected by:

106591

Vats in Office/Shop 8/09/90 1530 Bill Clark

TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag Methyl Ethyl Ketone	<2 <30 <2 <6 64 <0.18 15 <6 <200	<0.1 <1.5 <0.1 <0.3 3.2 <0.009 0.75 <0.3	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0 200
IGNITABILITY Closed Cup Flash Point	>150 ° F		140 ° F
REACTIVITY, mg/kg			
Reactive Cyanide as CN Reactive Sulfide as S	<0.10 <8		250 500
CORROSIVITY, standard unit	S		
рН	6.7		2.0-12.5
Polychlorinated Biphenyl Total Halogens as Cl Methyl Isobutyl Carbinol	<5 mg/kg 490 mg/kg <400 mg/kg		

This sample does not exhibit the characteristic of a hazardous waste. Methyl ethyl ketone analysis was performed on 9/23/90.

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600 SOUTH 25TH STREET P. O. BOX 30615 BILLINGS, MT 59107 (406) 248-9161 FAX (406) 248-9282

## TECHNICAL REPORT



EPORT TO:

CHEN-NORTHERN, INC. ATTN: MR. BILL CLARK

P O BOX 4699

HELENA, MT 59601

DATE: September 25, 1990

JOB NUMBER: 87-911

SHEET: 1 OF 4

INVOICE NO.: 104215

REPORT OF:

Solid Waste Analysis - Nellie Grant Mine

### Sample Identification:

On August 14, 1990, these samples of solid waste (laboratory numbers 106583, 106586, and 106592) were delivered to our laboratory for analysis. The tests were conducted in accordance with the U.S. Environmental Protection Agency Manual SW-846, <u>Test Methods for Evaluating Solid Waste</u>, 3rd Edition, November 1986 and 40 CFR Part 302 Method 1311 published June 29, 1990. The results of the analysis are shown on the following pages.

These samples could not be identified by field personnel as to their contents. Because they are clear, colorless to amber, nonviscous liquids, the tests include methyl ethyl ketone and methyl isobutyl carbinol. The toxicity characteristic for lab no. 106583 was determined in accordance with paragraph 1.2 of Method 1311. Other tests were performed to chemically identify the samples. These tests were selected based on the other materials believed to be present at the site.

A < sign indicates less than the reported value was present in the sample.

Reviewed by

rmr

September 25, 1990 Job No. 87-911 Sheet 2 of 4

Lab No.: Sample Description:

Date and Time Sampled:
Collected by:

106583 Barrel 2 8/09/90 1315

Bill Clark

TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag Methyl Ethyl Ketone	<2 <20 0.08 <3 12 <0.04 <3 <3 Total Volatiles <200	2 <20 0.08 <3 12 <0.04 <3 <3	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
IGNITABILITY Closed Cup Flash Point	>150	° F	140 ° F
REACTIVITY, mg/kg Reactive Cyanide as CN Reactive Sulfide as S	<0.1 <15	0	250 500
CORROSIVITY, standard units	12.9		2.0-12.5
Methyl Isobutyl Carbinol, % Total Solids at 105°C Total Sodium as Na, % Total Phosphorus as P, % Total Sulfur as S, % Total Kjeldahl Nitrogen, %	9.1 28.3 3.1 0.4 2.3 1.1	2	

Because this sample is a clear liquid containing no suspended solids, the TCLP concentrations represented are by the total metals.

This sample should be considered a hazardous waste due to its toxicity and corrosivity chacteristics.

September 25, 1990 Job No. 87-911 Sheet 3 of 4

Lab No.:
Sample Description:
Date and Time Sampled:
Collected by:

106586 Barrel 11 8/09/90 1345 Bill Clark

Collected by:	BILL	Clark	
TOXICITY		deasured	EPA Maximum for Non- Hazardous Waste mg/l
		ICLE, IIIY/I	ilių/ I
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag		<0.18 <0.1 0.008 <0.02 <0.02 <0.002 <0.18 <0.03	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
Methyl Ethyl Ketone		<200	200
IGNITABILITY			
Closed Cup Flash Point	115 ° F		140 ° F
REACTIVITY, mg/kg			
Reactive Cyanide as CN Reactive Sulfide as S	<0.10 <15		250 500
CORROSIVITY, standard units			
рН	2.4		2.0-12.5
Methyl Isobutyl Carbinol, % Total Solids at 105°C Total Sodium as Na, %	100 <1 0.01		

This sample exhibits the hazardous waste characteristics of ignitability.

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September 25, 1990 Job No. 87-911 Sheet 4 of 4

Lab No.:	106592
Sample Description:	Barrel 1
Date Sampled:	8/10/90
Collected by:	Bill Clark

corrected by:	DITT CTAIR	
TOXICITY	Managera	EPA Maximum for Non-
	Measured TCLP, mg/l	Hazardous Waste
Arsenic as As	<0.18	5.0
Barium as Ba	<0.1	100
Cadmium as Cd Chromium as Cr	<0.005 <0.02	1.0 5.0
Lead as Pb	<0.02	5.0
Mercury as Hg	<0.006	0.2
Selenium as Se	<0.18	1.0
Silver as Ag	<0.02	5.0
Methyl Ethyl Ketone	<200	200
IGNITABILITY		
Closed Cup Flash Point	111 ° F	140 ° F
REACTIVITY, mg/kg		
Reactive Cyanide as CN	<0.12	250
Reactive Sulfide as S	<9.3	500
CORROSIVITY, standard units		
рН	7.0	2.0-12.5
	-	2.0 12.0
Methyl Isobutyl Carbinol, %	93.3	
Total Solids at 105 °C	<1	
Total Sodium as Na, %	<0.01	

This sample exhibits the hazardous waste characteristics of ignitability.

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### TECHNICAL REPORT



REPORT TO:

CHEN-NORTHERN, INC. ATTN: MR. BILL CLARK

P 0 BOX 4699

HELENA, MT 59601

DATE: September 25, 1990

JOB NUMBER: 87-911 SHEET: 1 OF 3

INVOICE NO.: 104215

REPORT OF:

Solid Waste Analysis - Nellie Grant Mine

### Sample Identification:

On August 14, 1990, these samples of solid waste (laboratory numbers 106575 and 106576) were delivered to our laboratory for analysis. The tests were conducted in accordance with the U.S. Environmental Protection Agency Manual SW-846, <u>Test Methods for Evaluating Solid Waste</u>, 3rd Edition, November 1986 and 40 CFR Part 302 Method 1311 published June 29, 1990. The results of the analysis are shown on the following pages.

These samples were identified by field sampling personnel as possibly being anhydrous sodium sulfite. Based on this information, the analytical scheme was designed to include sodium, sulfur and sulfite tests to identify the sample.

Because of the possibility of reaction of the sample labeled Lab No. 106575 with acids to form toxic sulfur gases in the TCLP test, a total metals analysis was performed to determine toxicity in accordance with paragraph 1.2 of Method 1311. Theoretical concentrations were calculated based on the dilution factor of 20 required by the extraction.

A < sign indicates less than the reported value was present in the sample.

Reviewed by

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September 25, 1990 Job No. 87-911 Sheet 2 of 3

Lab No.:

Sample Description:

Date and Time Sampled:

Collected by:

106575

A Sacks Sodium Sulfite Anhydrous

8/09/90 1205 Bill Clark

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TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As	2	0.1	5.0
Barium as Ba	<20	<1	100
Cadmium as Cd	<0.9	<0.05	1.0
Chromium as Cr	<4	<0.2	5.0
Lead as Pb	14	0.7	5.0
Mercury as Hg	<1.0	<0.05	0.2
Selenium as Se	<3	<0.15	1.0
Silver as Ag	<4	<0.2	5.0
IGNITABILITY			
Closed Cup Flash Point	This sample is a solid that, when exposed to open flame, does not sustain combustion		140°F
REACTIVITY, mg/kg			
Reactive Cyanide as CN	<0.0	5	250
Reactive Sulfide as S	<1.0		500
CORROSIVITY, standard units			
рН	This sample is	a solid	2.0-12.5
Total Sodium as Na, % Total Sulfite as SO <sub>3</sub> , % Total Sulfur as S, % Total Solids at 105°C, %	38.2 47.5 27.0 100		

September 25, 1990 Job No. 87-911 Sheet 3 of 3

Lab No.:

Sample Description: Date and Time Sampled:

Collected by:

106576

Spill Adjacent to A 8/09/90 1210

Bill Clark

TOXICITY	Total Metals, mg/kg	Actual TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	71 <30 3 <6 230 <1.1 <5 <6	0.60 <0.4 0.22 <0.02 <0.07 <0.0005 <0.07 <0.15	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
IGNITABILITY Closed Cup Flash Point	This sample is a solid that, when exposed to open flame, does not sustain combustion		140 ° F
REACTIVITY, mg/kg			
Reactive Cyanide as CN Reactive Sulfide as S	<0.0 <1.0	_	250 500
CORROSIVITY, standard units			
рН	This sample is	a solid	2.0-12.5
Total Sodium as Na, % Total Sulfite as SO <sub>3</sub> , % Total Sulfur as S, % Total Solids at 105°C, %	28.2 22.0 18.2 74.9		

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## TECHNICAL REPORT



REPORT TO:

CHEN-NORTHERN, INC. ATTN: MR. BILL CLARK

P 0 BOX 4699

HELENA, MT 59601

September 25, 1990 DATE:

87-911 JOB NUMBER:

OF 104215 SHEET:

INVOICE NO .:

REPORT OF:

Solid Waste Analysis - Nellie Grant Mine

### Sample Identification:

On August 14, 1990, this sample of solid waste (laboratory number 106587) was delivered to our laboratory for analysis. The tests were conducted in accordance with the U.S. Environmental Protection Agency Manual SW-846, <u>Test Methods for</u> Evaluating Solid Waste, 3rd Edition, November 1986 and 40 CFR Part 302 Method 1311 published June 29, 1990. The results of the analysis are shown on the following page.

The field personnel identified the samples as possibly containing American Cyanamid SuperFloc 330. The tests were selected based on material safety data to chemically identify the sample contents. The toxicity was estimated by measuring the total metals concentration and calculating a TCLP extract concentration (paragraph 1.2 of Method 1311).

A < sign indicates less than the reported value was present in the sample.

Reviewed by Kathlen A.

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September 25, 1990 Job No. 87-911 Sheet 2 of 2

Lab No.:
Sample Description:
Date and Time Sampled:
Collected by:

106587 Barrel 12 8/09/90 1400 Bill Clark

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TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	<2 <20 <1 <4 33 <0.17 <3 <4	<0.1 <1 <0.05 <0.2 1.6 <0.0085 <0.15 <0.2	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
IGNITABILITY			-
Closed Cup Flash Point	>150 °	F	140 ° F
REACTIVITY, mg/kg			
Reactive Cyanide as CN Reactive Sulfide as S	<0.2 <20	5	250 500
CORROSIVITY, standard unit	ts		***************************************
рН	5.4		2.0-12.5
Total Solids at 105°C Total Sodium as Na, % Total Kjeldahl Nitrogen as	0.65	urately determi	ned

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600 SOUTH 25TH STREET P. O. BOX 30615 BILLINGS, MT 59107 (406) 248-9161 FAX (406) 248-9282

### TECHNICAL REPORT



EPORT TO:

CHEN-NORTHERN, INC. ATTN: MR. BILL CLARK

P O BOX 4699

HELENA, MT 59601

DATE: September 25, 1990

JOB NUMBER: 87-911

SHEET: 1 OF 3

EPORT OF:

Solid Waste Analysis - Nellie Grant Mine

### Sample Identification:

On August 14, 1990, these samples of solid waste (laboratory numbers 106588 and 106589) were delivered to our laboratory for analysis. The tests were conducted in accordance with the U.S. Environmental Protection Agency Manual SW-846, <u>Test Methods for Evaluating Solid Waste</u>, 3rd Edition, November 1986 and 40 CFR Part 302 Method 1311 published June 29, 1990. The results of the analysis are shown on the following pages.

These samples were identified by sample collection personnel as possibly being methyl ethyl ketone or methyl isobutyl carbinol. However, their black, viscous liquid appearance indicates they contain other materials possibly in addition to these. Tests were performed in order to chemically identify the contents of the samples. These were selected based on information from field personnel which identified materials that were possibly at the site.

A < sign indicates less than the reported value was present in the sample.

Reviewed by

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September 25, 1990 Job No. 87-911 Sheet 2 of 3

Lab No.: Sample Description: Date and Time Sampled: Collected by: 106588 Container B 8/09/90 1415 Bill Clark

corrected by:	DI I	II Clark	
TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	77 <20 <1 6 22 <4 <4 10	3.85 <1 <0.05 <0.3 1.1 <0.2 <0.2 0.5	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
Methyl Ethyl Ketone	Total Volatiles <200	<10	200
IGNITABILITY			
Closed Cup Flash Point	>150 ° F		140 ° F
REACTIVITY, mg/kg			
Reactive Cyanide as CN Reactive Sulfide as S	<0.10 <7.5		250 500
CORROSIVITY, standard uni	ts		
рН	7.1		2.0-12.5
Total Solids at 105°C Methyl Isobutyl Carbinol, Total Sodium as Na, % Total Sulfur as S, % Hydroxide as OH, % Total Phosphorus as P, % Total Kjeldahl Nitrogen a	mg/kg <500 <0.02 1.3 <0.01 4.15	urately determin	ned

September 25, 1990 Job No. 87-911 Sheet 3 of 3

Lab No.:
Sample Description:
Date Sampled:
Collected by.

106589 Container C 8/09/90 Bill Clark

TOXICITY		EPA Maximum
	Measured TCLP. mg/kg	for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	4 <20 <1 <4 22 <0.14 <4	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
Methyl Ethyl Ketone	<200	200
IGNITABILITY		
Closed Cup Flash Point	>150 ° F	140 ° F
REACTIVITY, mg/kg		
Reactive Cyanide as CN Reactive Sulfide as S	<0.10 33	250 250
CORROSIVITY, standard units		
На	1.4	2.0-12.5
Total Solids at 105°C  Methyl Isobutyl Carbinol, mg/kg  Total Sodium as Na, %  Total Sulfur as S, %  Hydroxide as OH, %  Total Phosphorus as P, %  Total Kjeldahl Nitrogen as N, %	Cannot be accurately of <400	determined

This sample exhibits the hazardous waste characteristics of corrosivity and toxicity.

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600 SOUTH 25TH STREET P. O. BOX 30615 BILLINGS, MT 59107 (406) 248-9161 FAX (406) 248-9282

### TECHNICAL REPORT



REPORT TO:

CHEN-NORTHERN, INC. ATTN: MR. BILL CLARK

P O BOX 4699

HELENA, MT 59601

DATE: September 25, 1990

JOB NUMBER: 87-911 SHEET: 1 OF 3 INVOICE NO.: 104215

REPORT OF:

Solid Waste Analysis - Nellie Grant Mine

#### Sample Identification:

On August 14, 1990, these samples of solid waste (laboratory numbers 106584 and 106585) were delivered to our laboratory for analysis. The tests were conducted in accordance with the U.S. Environmental Protection Agency Manual SW-846, <u>Test Methods for Evaluating Solid Waste</u>, 3rd Edition, November 1986 and 40 CFR Part 302 Method 1311 published June 29, 1990. The results of the analysis are shown on the following pages.

The field personnel identified the samples as possibly containing American Cyanamid Aerofloat 340 Promoter. The tests were performed to chemically identify the sample contents. These were selected based on material safety data provided by the American Cyanamid Company. The toxicity was estimated by measuring the total metals and calculating a theoretical TCLP extract concentration (paragraph 1.2 of Method 1311).

A < sign indicates less than the reported value was present in the sample.

Reviewed by

rmr

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September 25, 1990 Job No. 87-911 Sheet 2 of 3

Lab No.:

Sample Description: Date and Time Sampled: 106584 Barrel 3 8/09/90 1320 Bill Clark

Collected by:

TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	9 <30 <2 6 24 <0.11 <5 <6	0.45 <1.5 <0.1 <0.3 1.2 <0.0055 <0.25 <0.3	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
Methyl Ethyl Ketone	Total Volatiles <200	<10	200
IGNITABILITY			
Closed Cup Flash Point	100	° F	140 ° F
REACTIVITY, mg/kg			
Reactive Cyanide as CN Reactive Sulfide as S	<0.1 510	0	250 500
CORROSIVITY, standard unit	zs .		
рН	1.7		2.0-12.5
Total Solids at 105°C Total Sodium as Na, % Total Phosphorus as P, % Total Sulfur as S, % Total Kjeldahl Nitrogen as Methyl Isobutyl Carbinol,	<0.03 4.7 1.0 s N, %		ed

This sample exhibits the hazardous waste characteristics of ignitability, corrosivity and reactivity.

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September 25, 1990 Job No. 87-911 Sheet 3 of 3

Lab No.: Sample Description: Date and Time Sampled: Collected by: 106585 Barrel 4 8/09/90 1330 Bill Clark

corrected by:		II CIAIK	
TOXICITY	Total <u>Metals, mg/kg</u>	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	4 <20 <1 4 9 <0.08 <3 <4	4 <20 <1 4 9 <0.08 <3 <4	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
Methyl Ethyl Ketone	Total Volatiles <200	<200	200
IGNITABILITY			
Closed Cup Flash Point	>150 ° F	=	140°F
REACTIVITY, mg/kg ·	<del></del>		
Reactive Cyanide as CN Reactive Sulfide as S	<0.1 56	10	250 500
CORROSIVITY, standard un	its		
рН	10.0	)	2.0-12.5
Total Solids at 105°C Total Sodium as Na, % Total Phosphorus as P, % Total Sulfur as S, % Total Kjeldahl Nitrogen Methyl Isobutyl Carbinol	<0.03 3.93 1.1 as N, % 4.1		ned

### ien-Northern, Inc.

A member of the HIH group of companies

600 SOUTH 25TH STREET P. O. BOX 30615 BILLINGS, MT 59107 (406) 248-9161 FAX (406) 248-9282





REPORT TO:

CHEN-NORTHERN, INC. ATTN: MR. BILL CLARK

P 0 BOX 4699

HELENA, MT 59601

DATE: September 26, 1990

JOB NUMBER: 87-911

SHEET: 1 OF 2

INVOICE NO.: 104215

REPORT OF:

Solid Waste Analysis - Nellie Grant Mine

#### Sample Identification:

On August 14, 1990, these soil samples (laboratory numbers 106593 and 106594) were received in our laboratory for analysis. Total petroleum hydrocarbon determinations were made in accordance with Environmental Protection Agency Method 418.1. Polychlorinated biphenyls were determined by Energy Laboratories, Inc. of Billings, Montana.

The test results are shown on the following page. A < sign indicates less than the reported value was present in the sample.

Reviewed by

rmr

September 26, 1990 Job No. 87-911 Sheet 2 of 2

Lab No.:	106593 H Soil by	106594 I Soil by	
Sample Descriptin:	Office Shop	l Soil by Shed	
Date Sampled:	8/10/90	8/10/90	
Collected by:	Bill Clark	Bill Clark	
Total Recoverable Petroleum Hydrocarbons (418.1), mg/kg As Received: Dry Basis:	61,000 64,000	48,000 50,500	
Polychlorinated Biphenyls, m As Received: Dry Basis:	eg/kg <2.0 <2.1	<2.0 <2.1	
Moisture, %	5.3	5.0	

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### TECHNICAL REPORT



REPORT TO:

CHEN-NORTHERN, INC. ATTN: MR. BILL CLARK P 0 BOX 4699

HELENA, MT 59601

September 25, 1990 DATE:

87-911 JOB NUMBER: 1 OF SHEET: 104215

INVOICE NO .:

REPORT OF:

Solid Waste Analysis - Nellie Grant Mine

#### Sample Identification:

On August 14, 1990, these samples of solid waste (laboratory numbers 106577 -106582) were delivered to our laboratory for analysis. The tests were conducted in accordance with the U.S. Environmental Protection Agency Manual SW-846, Test Methods for Evaluating Solid Waste, 3rd Edition, November 1986 and 40 CFR Part 302 Method 1311 published June 29, 1990. The results of the analysis are shown on the following pages.

These samples were identified by field personnel as possibly containing sodium isopropyl xanthate. All six samples consisted of yellow to white, mottled, dry pellets. Because material safety data for sodium isopropyl xanthate indicated that under heated conditions or when exposed to acids, this material would liberate toxic sulfur gases, the toxicity was estimated by measuring total metals and calculating a theoretical TCLP extract concentration. (Paragraph 1.2 of Method 1311.)

A < sign indicates less than the reported value was present in the sample.

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September 25, 1990 Job No. 87-911 Sheet 2 of 7

Lab No.: Sample Description: Date and Time Sampled: Collected by: 106577 Barrel 5 8/09/90 1235 Bill Clark

TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	<2 <40 <2 <7 <7 <1.0 8 <7	<0.1 <2 <0.1 <0.35 <0.35 <0.05 0.4	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
IGNITABILITY Clased Cup Flash Point	This sample is	a solid and	140 ° F
Closed Cup Flash Point	when exposed t does not susta	o open flame,	140 F
REACTIVITY, mg/kg			
Reactive Cyanide as CN Reactive Sulfide as S	<0.0 3600	5	250 500
CORROSIVITY, standard units			
рН	This sample is	a solid	2.0-12.5
Total Sodium as Na, % Total Sulfur as S, % Total Sulfide as S, % Total Hydroxide as OH, % Total Solids at 105°C	15 11.5 7.3 0.4 56.8	4	

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September 25, 1990 Job No. 87-911 Sheet 3 of 7

Lab No.:
Sample Description:
Date and Time Sampled:
Collected by:

106578 Barrel 6 8/09/90 1240 Bill Clark

Collected by:	B1	II Clark	
TOXICITY	Total	Theoretical	EPA Maximum for Non- Hazardous Waste
	Metals, mg/kg	TCLP, mg/l	mg/1
Arsenic as As	<2	<0.1	5.0
Barium as Ba	<40	<2	100
Cadmium as Cd	<2	<0.1	1.0
Chromium as Cr	<8 40	<0.4	5.0
Lead as Pb	<8	<0.4	5.0
Mercury as Hg	<1.0	<0.05	0.2
Selenium as Se	<6 <8	<0.3	1.0
Silver as Ag	<8	<0.4	5.0
IGNITABILITY			
Closed Cup Flash Point	This sample is when exposed t does not susta	140 ° F	
REACTIVITY, mg/kg			
Reactive Cyanide as CN	<0.0	5	250
Reactive Sulfide as S	2600		500
CORROSIVITY, standard units			
рН	This sample is	a solid	2.0-12.5
μπ	11113 3 amp 16 13	a Julia	2.0-12.3
Total Sodium as Na, %	14.4		
Total Sulfur as S, %	23.5		
Total Sulfide as S, %	7.8		
Total Hydroxide as OH, %	0.9		
Total Solids at 105 ° C	52.3		
10001 001100 00 100 0	02.3		

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September 25, 1990 Job No. 87-911 Sheet 4 of 7

Lab No.: Sample Description: Date and Time Sampled: Collected by: 106579 Barrel 7 8/09/90 1244 Bill Clark

TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l	
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	<2 <30 <2 <7 <7 <1.2 <5 <7	<0.1 <1.5 <0.1 <0.35 <0.35 <0.06 <0.25 <0.35	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0	
IGNITABILITY Closed Cup Flash Point	This sample is a solid and when exposed to open flame, does not sustain combustion		140 ° F	
REACTIVITY, mg/kg Reactive Cyanide as CN Reactive Sulfide as S	<0.0 7700	5	250 500	
CORROSIVITY, standard units pH	This sample is	a solid	2.0-12.5	
Total Sodium as Na, % Total Sulfur as S, % Total Sulfide as S, % Total Hydroxide as OH, % Total Solids at 105 ° C	14.1 19.0 3.9 0.3 43.0	1		

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September 25, 1990 Job No. 87-911 Sheet 5 of 7

Lab No.: Sample Description: Date and Time Sampled: Collected by: 106580 Barrel 8 8/09/90 1248 Bill Clark

TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l	
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	<2 <20 <1 <4 <4 <1.0 <6 <4	<0.1 <1 <0.05 <0.2 <0.2 <0.05 0.3 <0.2	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0	
IGNITABILITY				
Closed Cup Flash Point	This sample is when exposed to does not susta	140 ° F		
REACTIVITY, mg/kg				
Reactive Cyanide as CN Reactive Sulfide as S	<0.0 7100	5	250 500	
CORROSIVITY, standard units				
рН	This sample is	a solid	2.0-12.5	
Total Sodium as Na, % Total Sulfur as S, % Total Sulfide as S, % Total Hydroxide as OH, % Total Solids at 105°C	16.2 15.1 4.0 0.3 44.4	5		

September 25, 1990 Job No. 87-911 Sheet 6 of 7

Lab No.: Sample Description: Date and Time Sampled: 106581 Barrel 9 8/09/90 1250

Date and Time Sampled.	0/03/30	
Collected by:	Bill	Clark

TOXICITY	Total <u>Metals, mg/kg</u>	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	<2 <20 <1 <5 8 <1.2 <4 <5	<0.1 <1 <0.05 <0.25 0.4 <0.06 <0.2 <0.25	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0
IGNITABILITY Closed Cup Flash Point	This sample is when exposed to does not susta	o open flame,	140°F
REACTIVITY, mg/kg Reactive Cyanide as CN Reactive Sulfide as S	<0.0! 2200	5	250 500
CORROSIVITY, standard units pH	This sample is	a solid	2.0-12.5
Total Sodium as Na, % Total Sulfur as S, % Total Sulfide as S, % Total Hydroxide as OH, % Total Solids at 105°C	13.8 12.5 3.4 0.1 51.8		

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September 25, 1990 Job No. 87-911 Sheet 7 of 7

Lab No.:
Sample Description:
Date and Time Sampled:
Collected by:

106582 Barrel 10 8/09/90 1252 Bill Clark

TOXICITY	Total Metals, mg/kg	Theoretical TCLP, mg/l	EPA Maximum for Non- Hazardous Waste mg/l	
Arsenic as As Barium as Ba Cadmium as Cd Chromium as Cr Lead as Pb Mercury as Hg Selenium as Se Silver as Ag	<2 <20 <0.8 <3 12 <1.0 <4	<0.1 <1 <0.04 <0.06 0.6 <0.05 <0.2 <0.2	5.0 100 1.0 5.0 5.0 0.2 1.0 5.0	
IGNITABILITY Closed Cup Flash Point	This sample is when exposed t does not susta	o open flame,	140°F	
REACTIVITY, mg/kg				
Reactive Cyanide as CN Reactive Sulfide as S	<0.1 4300	0	250 500	
CORROSIVITY, standard units				
рН	This sample is	a solid	2.0-12.5	
Total Sodium as Na, % Total Sulfur as S, % Total Sulfide as S, % Total Hydroxide as OH, % Total Solids at 105°C	13.5 14.0 4.1 0.3	0		

#### APPENDIX C

PERSONS CONTACTED

#### PERSONS CONTACTED

Mr. Tom Whites, Montana Tunnels, Jefferson City, Montana, 933-8314

Ms. Kathy Roos, Special Resources Management, Butte, Montana, 494-2500

Ms. Nora Bessler, Montana Oil Processing, Great Falls, Montana, 761-4503

Ms. Pat Bond, American Cyanamid, 801-298-9381

Mr. Ray Tilman, Montana Resources Inc., Butte, Montana, 723-4081

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